O.15 Are plane-waves basis packages able to predict the correct long range behavior of the local kinetic energy per particle?

Camilo Espejo¹, Hugo J. Bohórquez² and Aldo H. Romero³

The Kinetic energy density $\tau(\mathbf{r})$ is an important quantity not only in density functional theory but also in wave-function methods used to compute the electronic structure of molecules and condensed matter systems. In particular, for finite systems it has been shown that $G = \tau(\mathbf{r})/n(\mathbf{r})$, which gives the local kinetic energy per particle, approaches asymptotically to the first ionization energy [1]. Such behavior is mostly due to the exponential decay of the multielectronic wave function as $r \to \infty$ [2]. It also represents an alternative to Koopmans' theorem for the calculation of the ionization energy in DFT [3]. In meta-GGA functionals τ is of prime importance since the XC energy depends explicitly on it. We propose to use this quantity as a quality indicator of the approximations and numerical implementations aiming to describe correctly the electronic properties of finite systems. In this work we have evaluated the asymptotic behavior of G in ABINIT for several rare gas atoms, Hydrogen and second raw elements using normconserving pseudopotentials and compare to Coupled Cluster and existing analytic solutions of the same problem. In the case of Hydrogen, the self-interaction error prevents an accurate DFT description. However, by using the Fermi-Amaldi correction and the bare coulomb potential of the proton, G approaches the exact value of the ionization energy in a systematic way. Our results show that for the case of finite systems, the common convergence studies of the total energy with respect to both plane-waves cut-off energy and unit cell size, is not enough to obtain the expected behavior of G in real space.



¹ Departamento de Física y Geociencias, Universidad del Norte, Barranquilla, Colombia.

² CienCo, Boqotá, Colombia.

³ Physics and Astronomy Department, West Virginia University, USA.

^[1] Zhong-Zhi Yang, Shubin Liu, and Yan Alexander Wang. Chem. Phys. Lett. 258, 30-36 (1996).

^[2] J. Katriel, and E. R. Davidson. Proc. Nat. Acad. Sci. 77 (8), 4403-4406 (1980).

^[3] Vanfleteren D, Van Neck D, Ayers PW, Morrison RC, and Bultinck P. J. Chem. Phys. **130**(19), 194104 (2009).